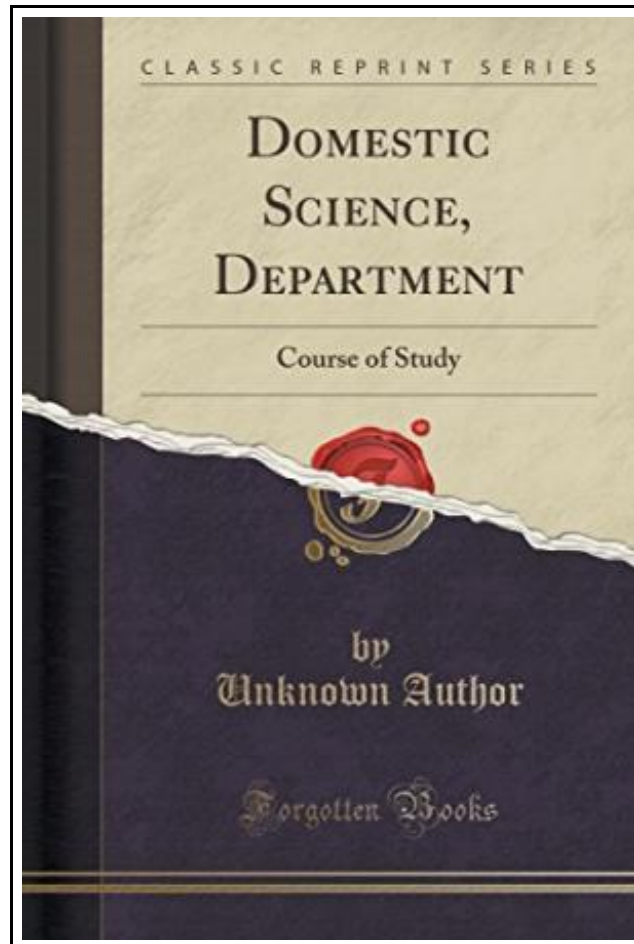


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Forgotten Books, United States, 2015. Paperback. Book Condition: New. 229 x 152 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.Excerpt from Domestic Science, Department: Course of Study Many substances that we are familiar with undergo changes. The burning candles change from an opaque solid to a translucent liquid. Salt, being mixed with water, becomes a clear liquid not distinguishable from water itself. Heat changes ice into water and water into steam. The burning of wood, which is the uniting of the solid carbon of the wood with the gas, oxygen, form a gas carbon dioxide, which is quite different from oxygen. When the action is over, a handful of gray ashes is all of the solid substance left. Some of these changes are only a change in the form of matter. The liquid candle grease will return to solid form when cooled; dissolved salt may be recovered by evaporating the water; steam may be collected and condensed into form of water again. These are called physical changes. That part of the melted candle which was burnt will not return to its original form. Burnt sugar will never be sweet and white again. The ashes can not be changed into wood. These changes are changes in the composition of matter, and are called chemical changes. Elements and Compounds: - Some substances consist of but one thing, hence are called simple, such as iron, oxygen and carbon. A simple substance is called an element. Some substances are composed of two or more elements, as water What two elements make up water? A substance composed of two or more elements is called a compound. In a mixture each substance keeps its own properties; in a compound these give place to new properties belonging to the compound. The most common...



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